

Exotic Archaeology: Searching for Superheavy Elements in Nature and Dating of Human DNA with the ^{14}C Bomb Peak

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of Particle Physics, Cosmology and Astrophysics
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Searching for Superheavy Elements in Nature

Early History of Superheavy Elements (SHE)

In the late 1960s, shell model corrections of the liquid drop model led to a region of stability for superheavy elements: Myers & Swiatecky, Nilsson, Strutinsky, Nix, Möller,

Number of protons

114

Island of superheavy elements (SHE) →

82

Sea of instability

50

Sn

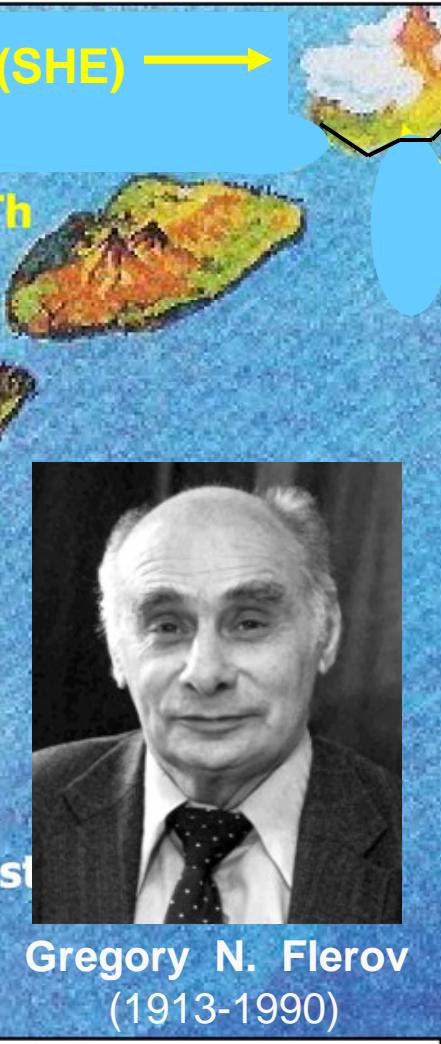
20

Ca

Pb

U, Th

Sea of insta



Gregory N. Flerov
(1913-1990)

20

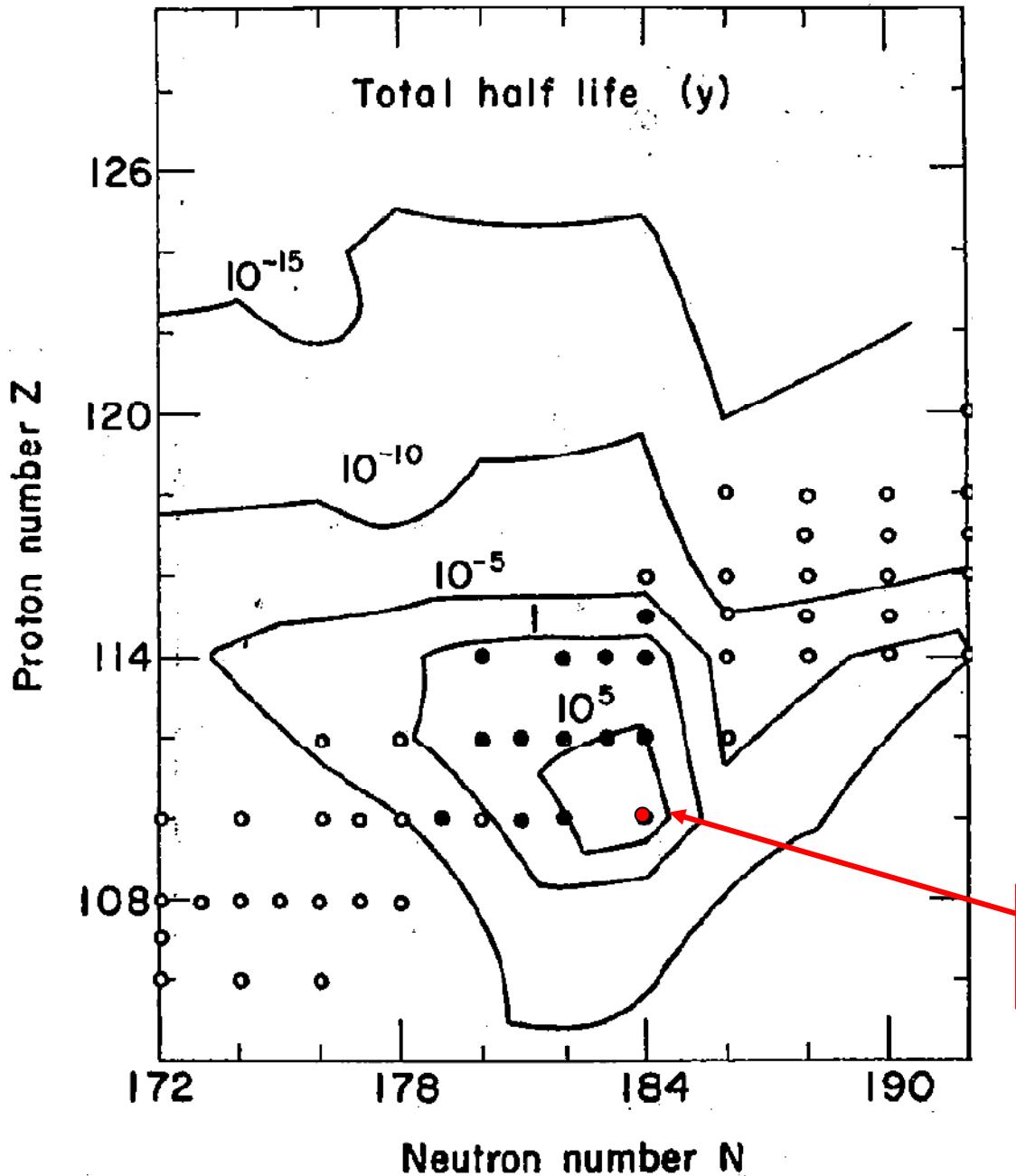
82

126

184

Number of neutrons

E. O. Fiset and J. R. Nix
Calculation of Half-Lives for Superheavy Nuclei
Nucl. Phys. A 193 (1972) 647- 671



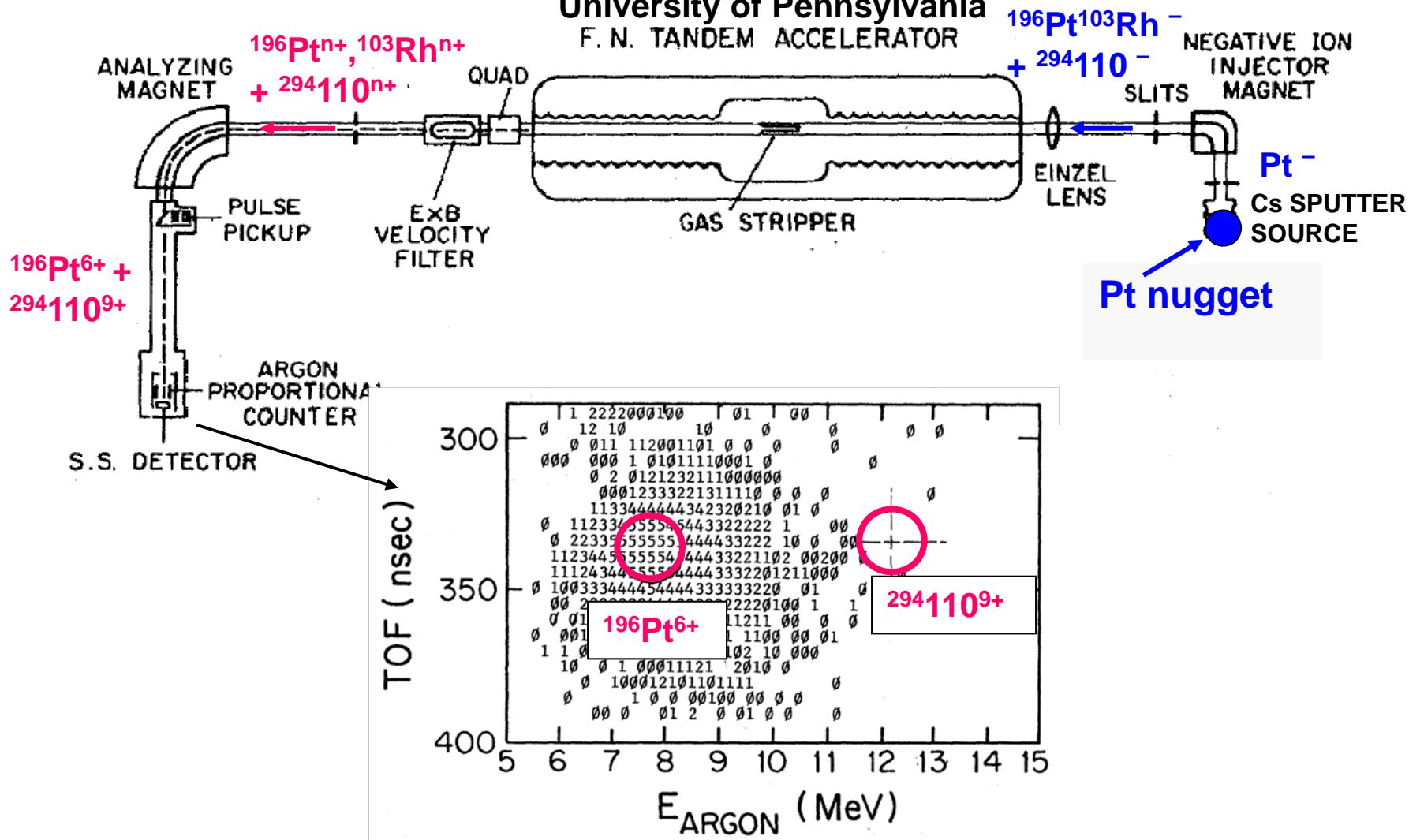
Contour plot of calculated total half-life for even superheavy nuclei. The solid points indicate β -stable nuclei whose total half-life is longer than 1 y, and the open circles β -stable nuclei whose total half-life is shorter than 1 y.

294110, Eka-Pt
 $t_{1/2} = 2.5 \times 10^9$ y

Early Search for the Superheavy Nuclide $^{294}\text{110}$ in a Natural Platinum Nugget with Accelerator Mass Spectrometry

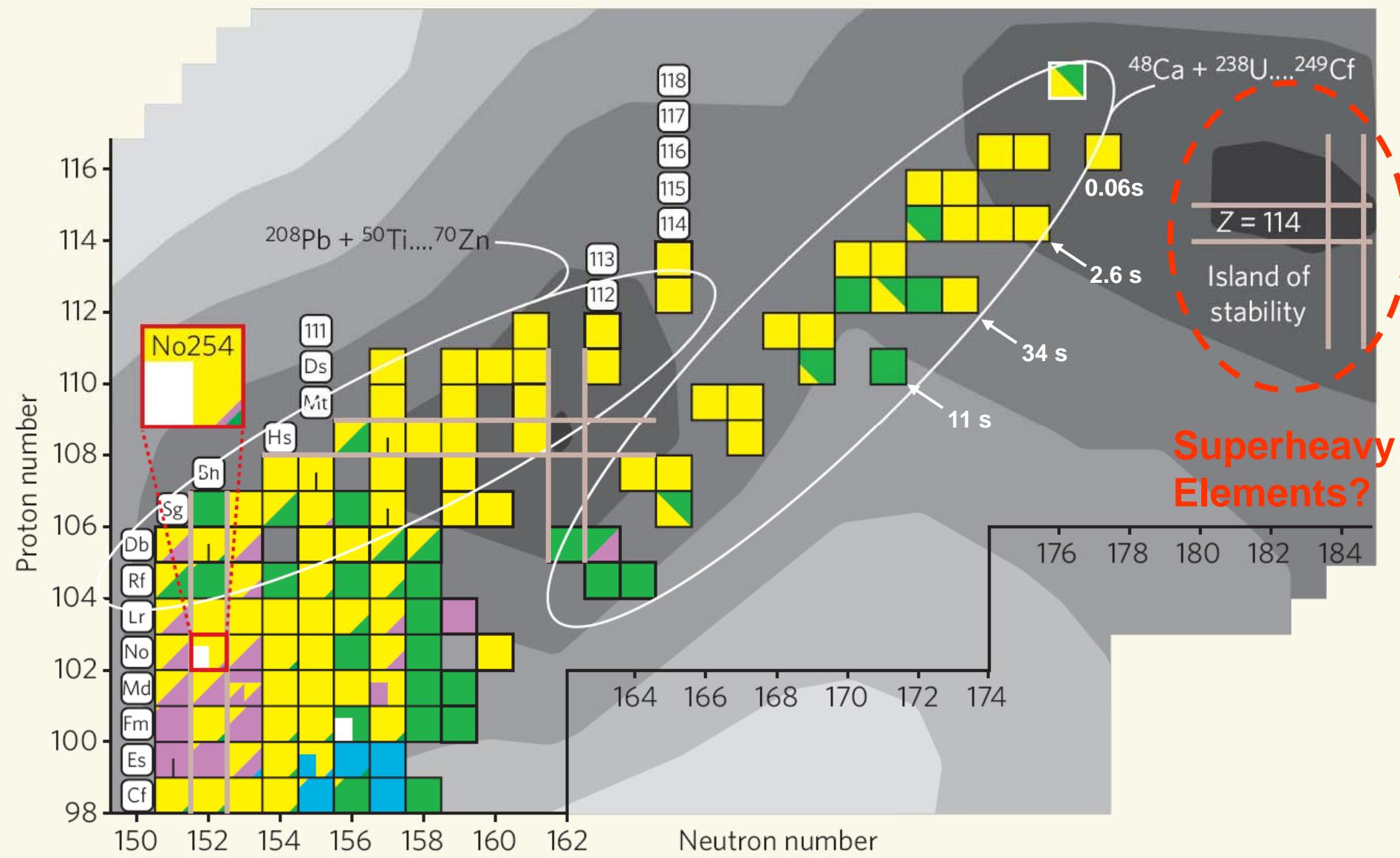
W. Stephens et al., Phys. Rev. C 21 (1980) 1964

University of Pennsylvania
F. N. TANDEM ACCELERATOR



Result: $294\text{110}/\text{Pt} < 10^{-11}$

Conclusion: $t_{1/2} < 1.7 \times 10^8$ yr, if $294\text{110}/\text{Pt} \sim 0.03$ in *Synthesis of SHE in r-Process*, Schramm and Fowler, Nature 231 (1971) 103



M.A. Stoyer, *Island ahoy!*, Nature 442 (2006) 876

A. Sobiczewski & K. Pomorski, Prog. Part. Nucl. Phys. 58 (2007) 292-349

**Superheavy
Elements?**

**Recent Searches for SHE in Nature with ICP-MS
(Inductively Coupled Plasma Mass Spectrometry)
by A. Marinov et al. (Hebrew University of Jerusalem)
report the possible existence at abundance levels
of 10^{-12} to 10^{-10} .**

A. Marinov, I. Rodushkin, Y. Kashiv et al.

Existence of long-lived isomeric states in naturally-occurring neutron-deficient Th isotopes

Phys. Rev. C76 (2007) 021303(R).

A. Marinov, I. Rodushkin, Y. Kashiv et al.

Evidence for a long-lived superheavy nucleus with atomic mass number A=292 and atomic number Z~122 in natural Th

arXiv: 0804.3869 (April 2008).

A. Marinov, I. Rodushkin, A. Pape et al.

Existence of long-lived isotopes of a superheavy element in natural Au

arXiv: nucl-ex/0702051v1 (February 2007).

Inductively Coupled Plasma-Sector Field Mass Spectrometer (ICP-SFMS)

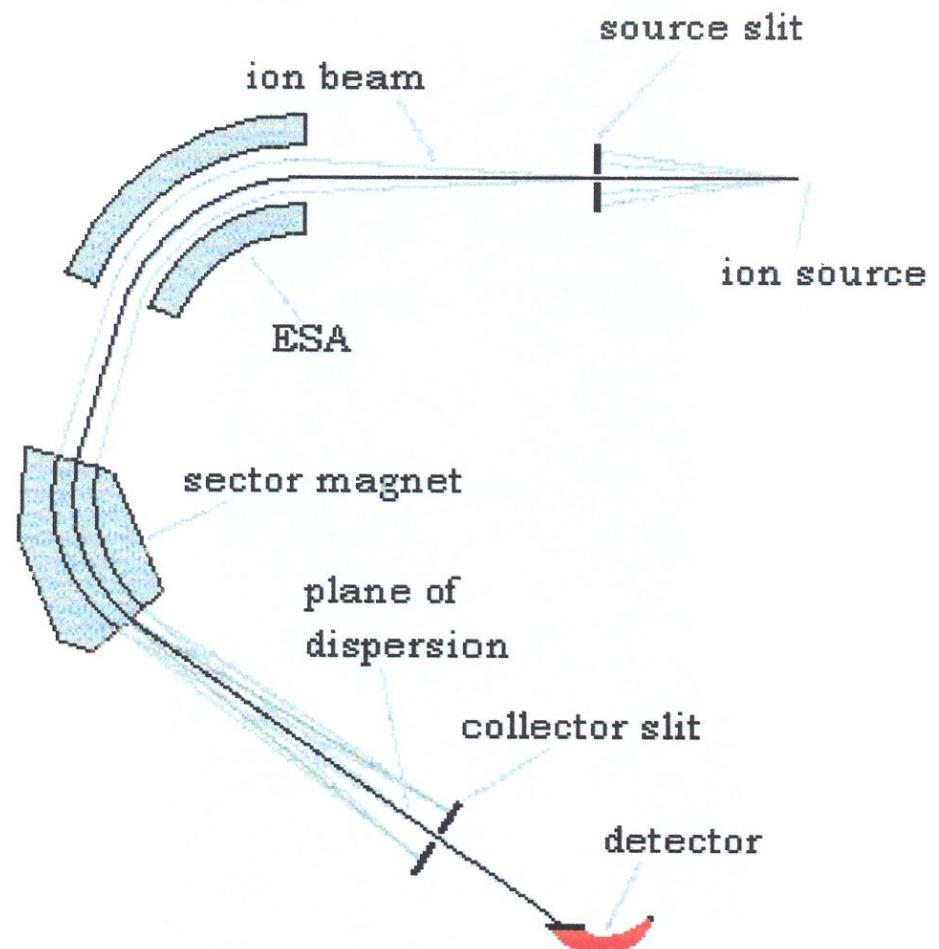
Plasma source at
6000-8000 K

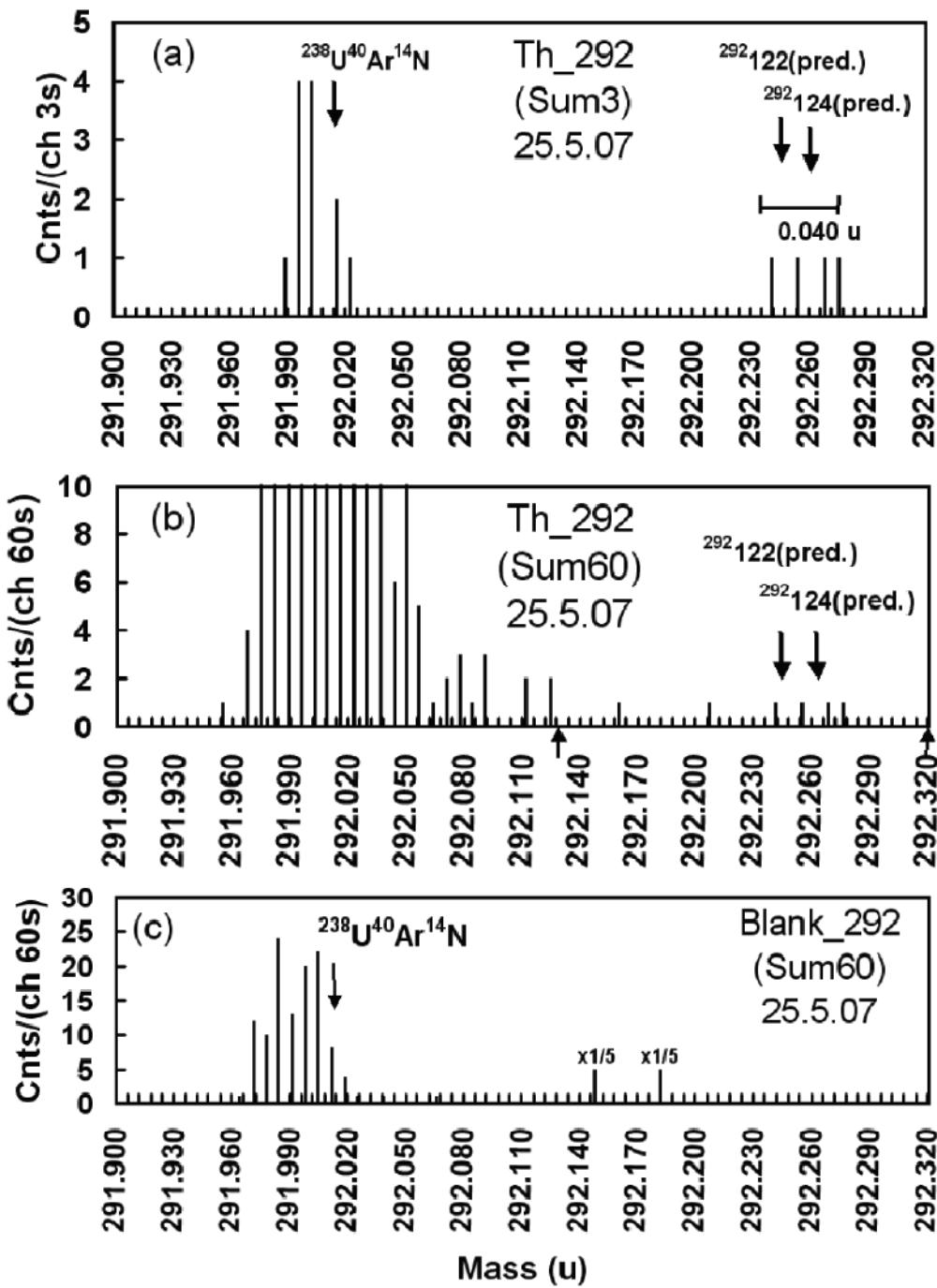
Mostly **atoms**
from the source

Studied material:
Thorium Solution

$$M/\Delta M = 4000$$

Courtesy of A. Marinov

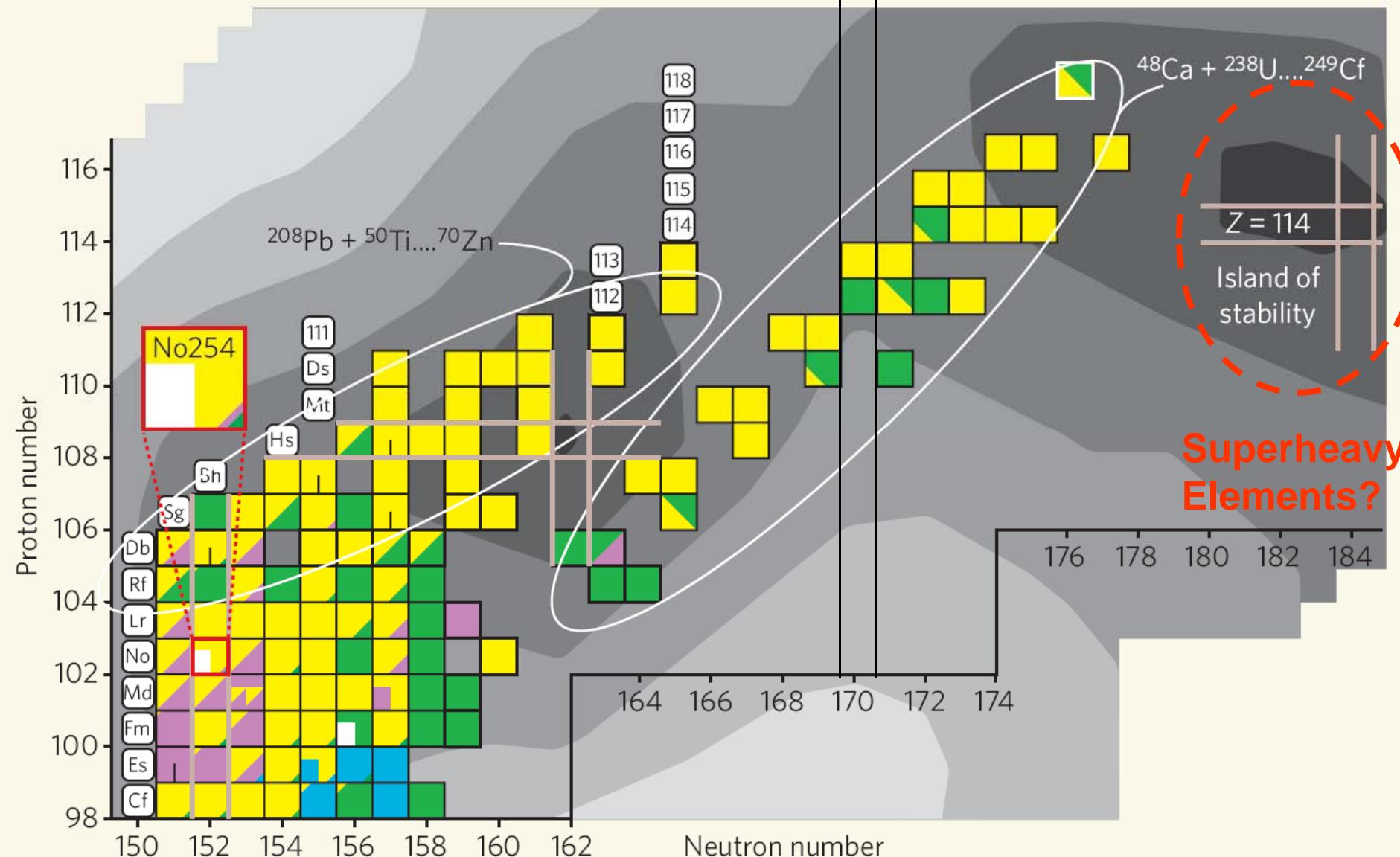




Marinov et al., “Evidence for a long-lived superheavy nucleus with atomic mass number A=292 and atomic number Z~122 in natural Th“, arXiv: 0804.3869 (April 2008)

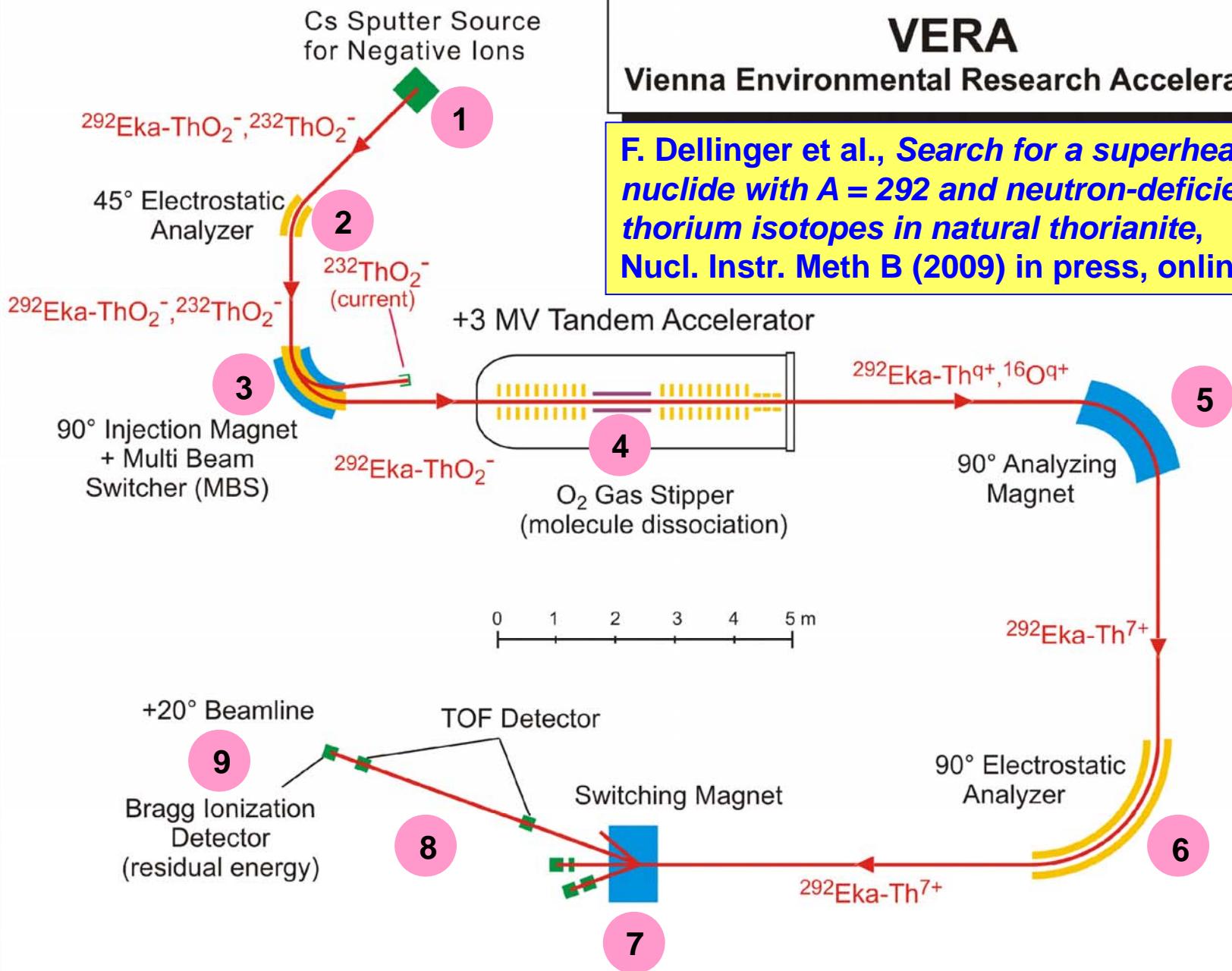
$$^{292}\text{Eka-Th}/^{232}\text{Th} = (1-10) \times 10^{-12}$$

122

²⁹²Eka-Th

VERA

Vienna Environmental Research Accelerator



F. Dellinger et al., *Search for a superheavy nuclide with A = 292 and neutron-deficient thorium isotopes in natural thorianite*, Nucl. Instr. Meth B (2009) in press, online

9

residual energy (MeV)

16

12

8

4

670

680

690

700

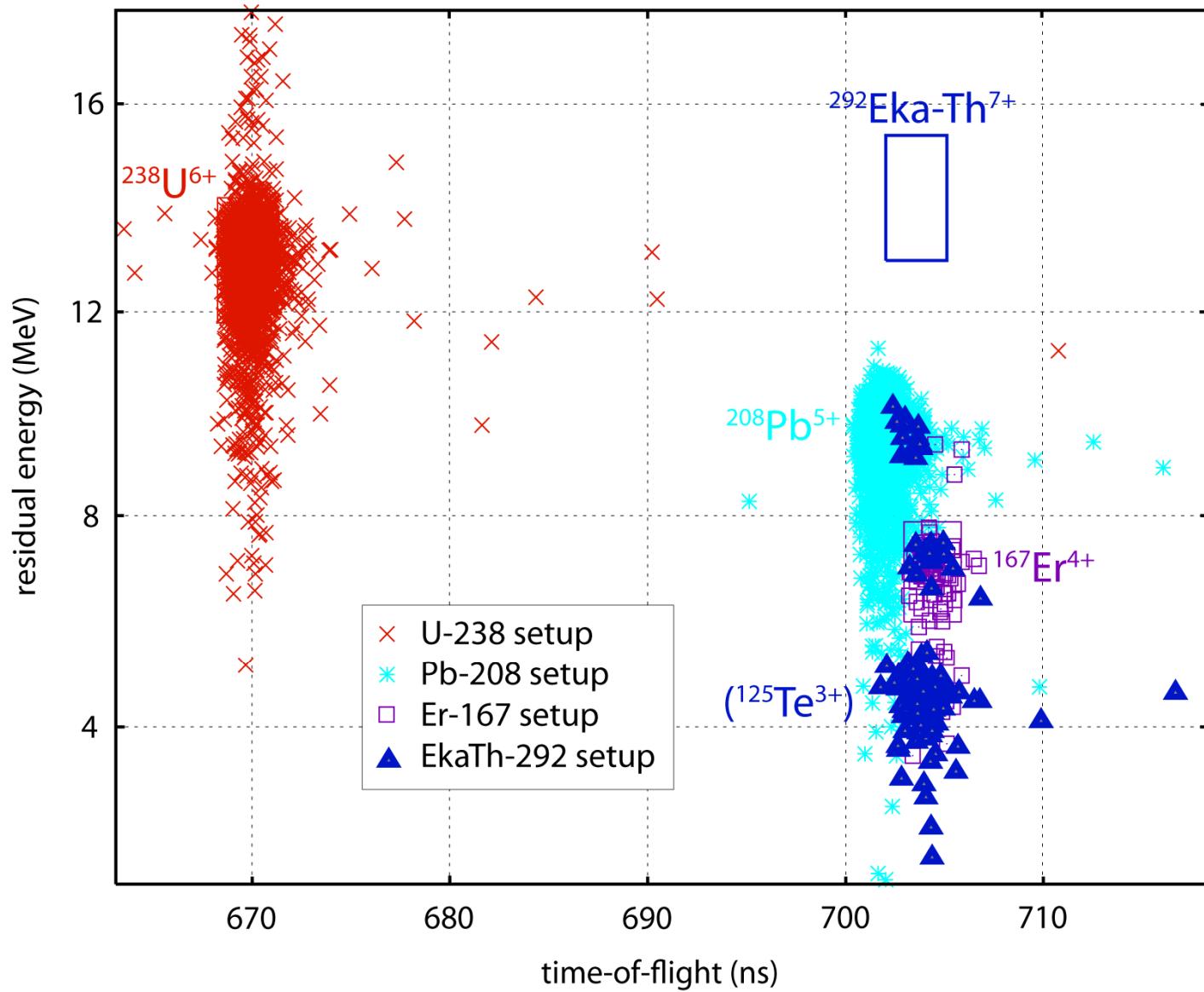
710

time-of-flight (ns)

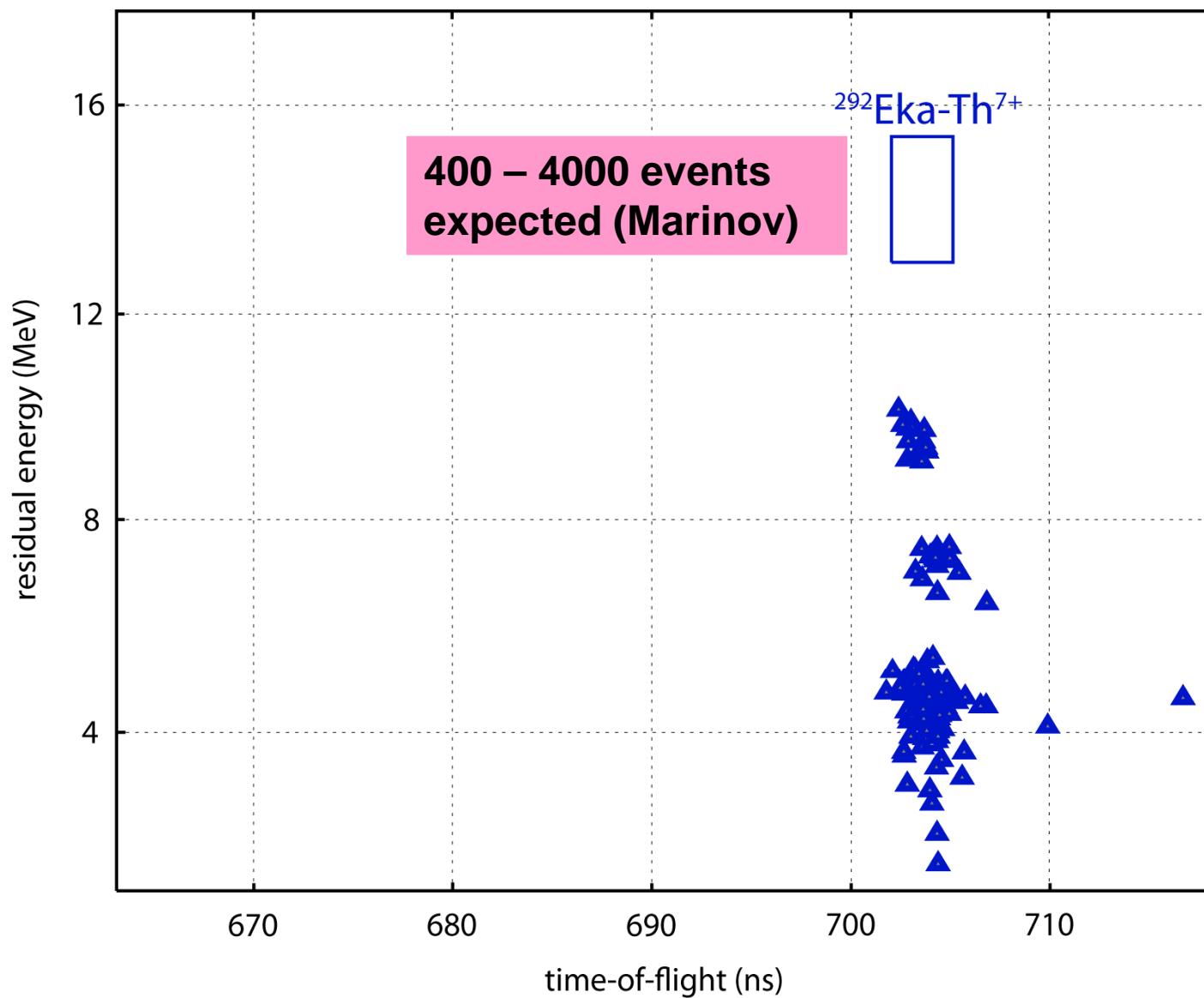
 $^{292}\text{Eka-Th}^{7+}$ $^{208}\text{Pb}^{5+}$ $^{167}\text{Er}^{4+}$ $(^{125}\text{Te}^{3+})$

8

Search for superheavy element $^{292}\text{Eka-Th}$



Search for superheavy element $^{292}\text{Eka-Th}$



Abundance of the rare isotopes relative to ^{232}Th

Rare isotope	Events detected	Counting time (h)	$^{232}\text{ThO}_2^-$ (mC)	Abundance rel. to ^{232}Th		
				AMS	Vienna ^a	ICP-SFMS
$^{292}\text{Eka-Th}$	0	31.5	36	<4x10 ⁻¹⁵	not meas.	(1-10)x10 ⁻¹²
^{211}Th	0	6.6	12	<5x10 ⁻¹⁵	<9.6x10 ⁻¹³	(1-10)x10 ⁻¹¹
^{213}Th	2	10.3	16	7x10 ⁻¹⁶ - 8x10 ⁻¹⁵	<1.2x10 ⁻¹²	(1-10)x10 ⁻¹¹
^{217}Th	1	10.3	16	1x10 ⁻¹⁶ - 6x10 ⁻¹⁵	<6.6x10 ⁻¹³	(1-10)x10 ⁻¹¹
^{218}Th	0	6.6	12	<5x10 ⁻¹⁵	<2.4x10 ⁻¹²	(1-10)x10 ⁻¹¹

^aDellinger et al., Nucl. Instr. Meth. B (2009) in press, online

^bLachner et al., Phys. Rev C78 (2008) 064313

^cMarinov et al., Phys. Rev. C76 (2007) 021303(R), $^{211,231,217,218}\text{Th}$; (April 2008), $^{292}\text{Eka-Th}$

H	2															He	2	
Li	Be																	
Na	Mg	3	4	5	6	7	8	9	10	11	12		B	C	N	O	F	Ne
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Al	Si	P	S	Cl	Ar	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
Cs	Ba	La*	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn	
Fr	Ra	Ac ⁺	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	₁₁₂ Cn	113	114	115	116		118	
119	120	121	154	155	156	157	158	159	160	161	162	163	164	165	166	167	168	

* Lanthanides

Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
58	59	60	61	62	63	64	65	66	67	68	69	70	71

+ Actinides

Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
90	91	92	93	94	95	96	97	98	99	100	101	102	103

Superactinides
(Seaborg 1969)

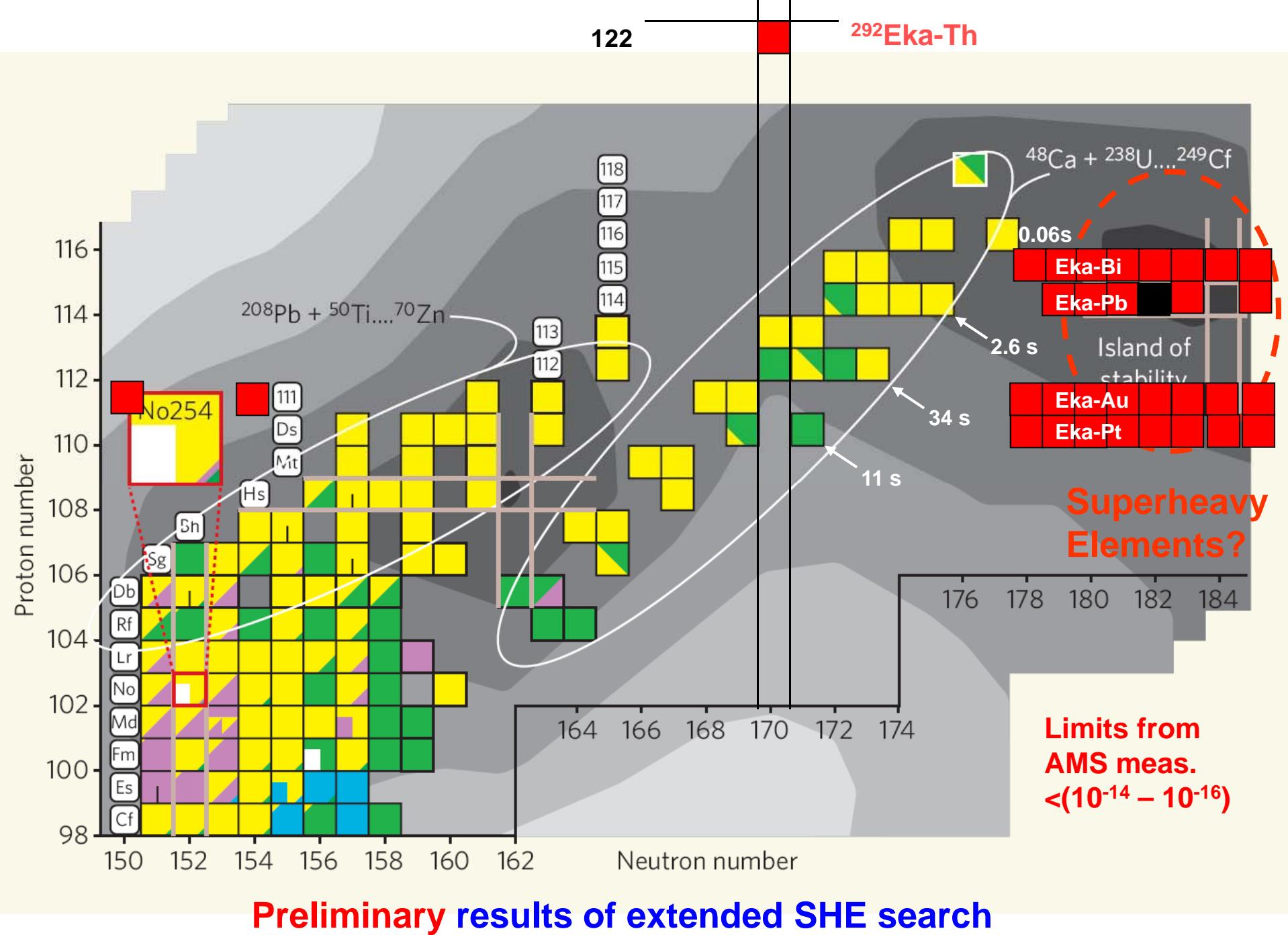


122 → 153

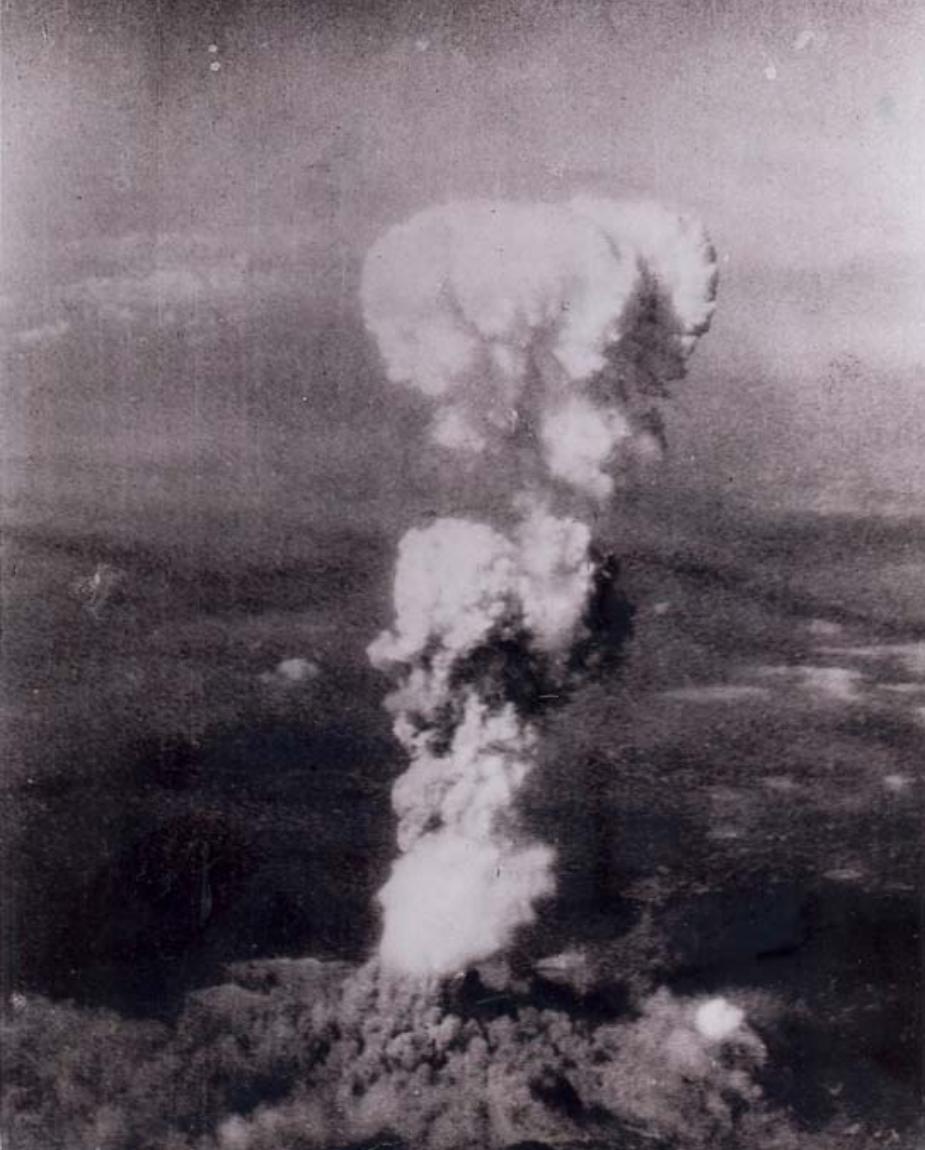
Modified figure from:
S. Hofmann, Lecture Notes in Physics,
Vol. 3, Springer, Heidelberg (2008)

Materials investigated for searches of SHE at VERA

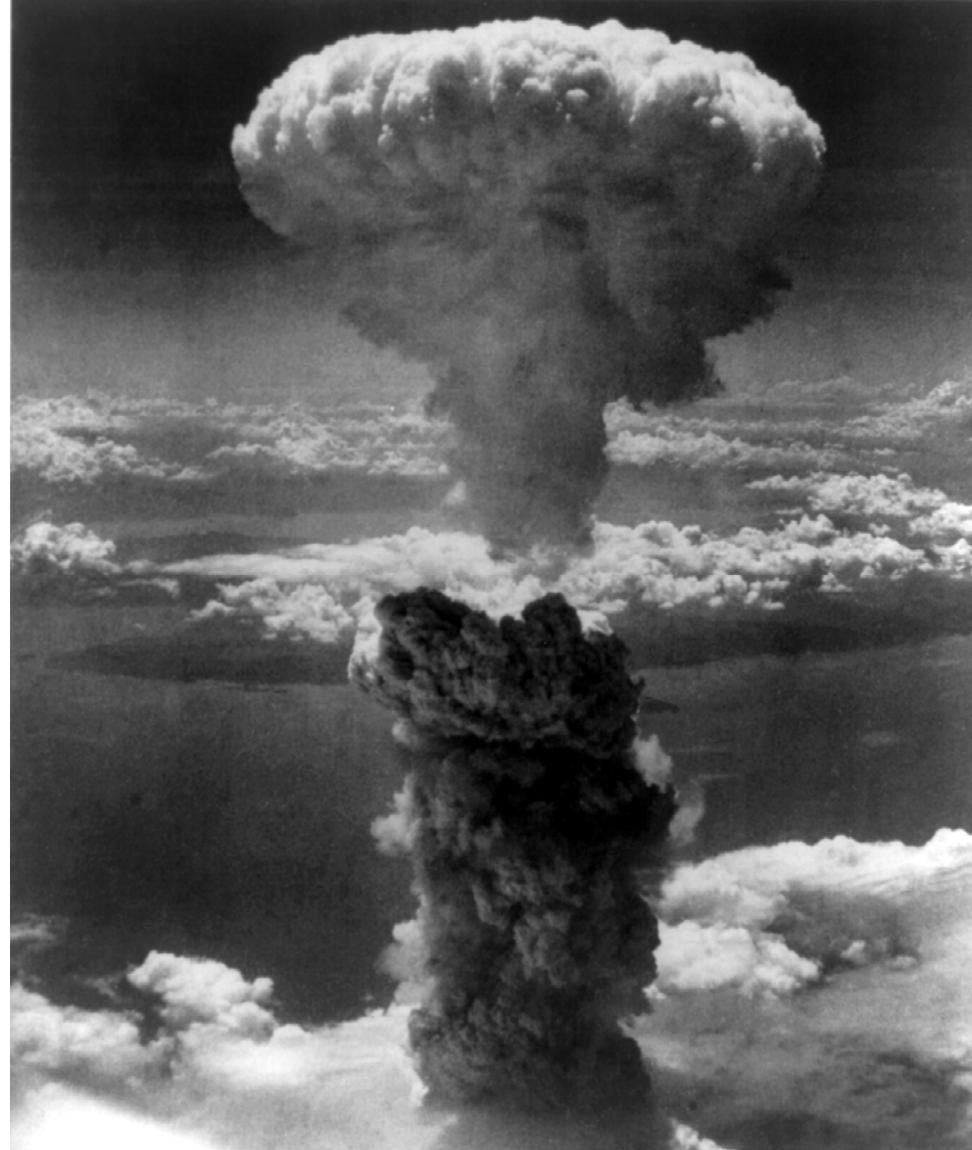
Element	eka-Elem.	Mass (amu)	Materials
Platin	110 (Ds)	planned	Nuggets of different origin
Gold	111 (Rg)	261, 265 289, 290, 291, 292 293, 294, 295, 296	Nuggets of different origin
Mercury	112	planned	Cinnabarit (HgS), neg. ions difficult
Thallium	113	planned	Avicennite (Tl_2O_3), Carlenite(Tl_2S) Lorandite ($TlAsS_2$)
Lead	114	295, 297, 299	Galenit (PbS) -> PbF_2 -> PbF_3^-
Bismuth	115	293, 294, 295, 296 297, 298, 299, 300	Bi(Alfa), Bismuth Ochre (Bi_2O_3)
Thorium	90 122	211, 213, 217, 218 292	Thorianite (ThO_2), commercial Thorium Oxide, Marinov Oxide



The Ugly and the Beautiful: Utilizing the ^{14}C Bomb Peak



The mushroom cloud over Hiroshima
after the dropping of *Little Boy* (U bomb)
6-Aug-1945



The mushroom cloud over Nagasaki
after the dropping of *Fat Man* (Pu bomb)
9-Aug-1945



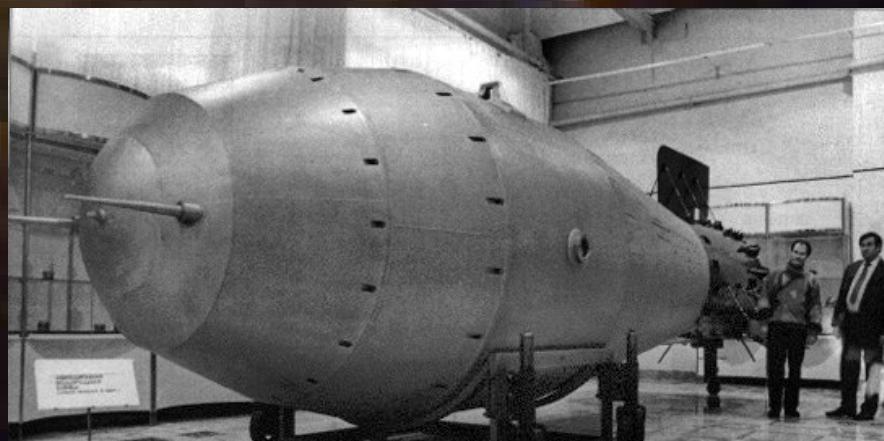
**Mike H-bomb test, 10.4 Megaton
(~700 Hiroshima bombs)
Enewetak Atoll, 1 Nov1952**



Bravo H-bomb test, 15 Megaton
(~1000 Hiroshima bombs)
Bikini Atoll, 1 March 1954



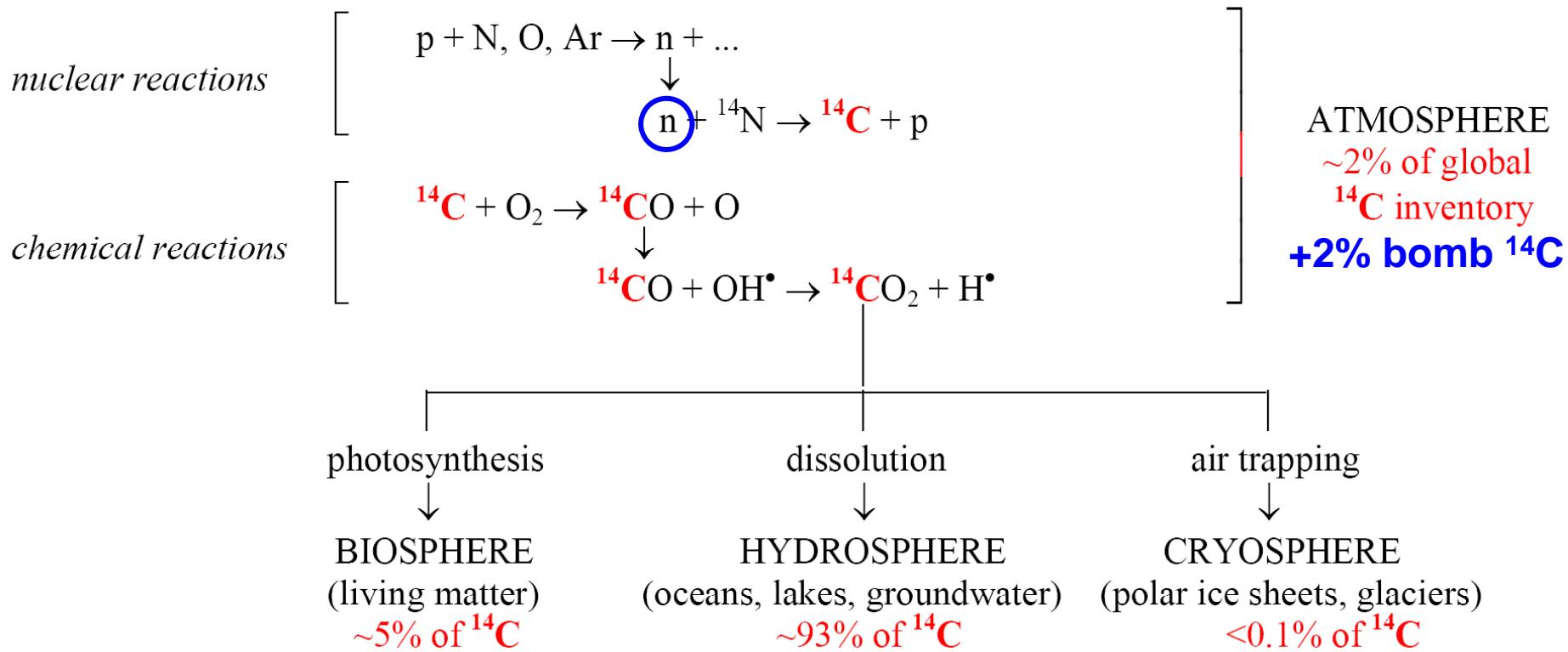
Tsar H-bomb test, 50 Megaton
(~3500 Hiroshima bombs)
Novaya Zemlya, 30 Oct 1961





John F. Kennedy signing the Nuclear Test Ban Treaty on 7 October 1963 in the White House, Washington D.C. (6 weeks later he was assassinated in Dallas, Texas)

Cosmic ray production and distribution of ^{14}C (half-life = 5730 years)



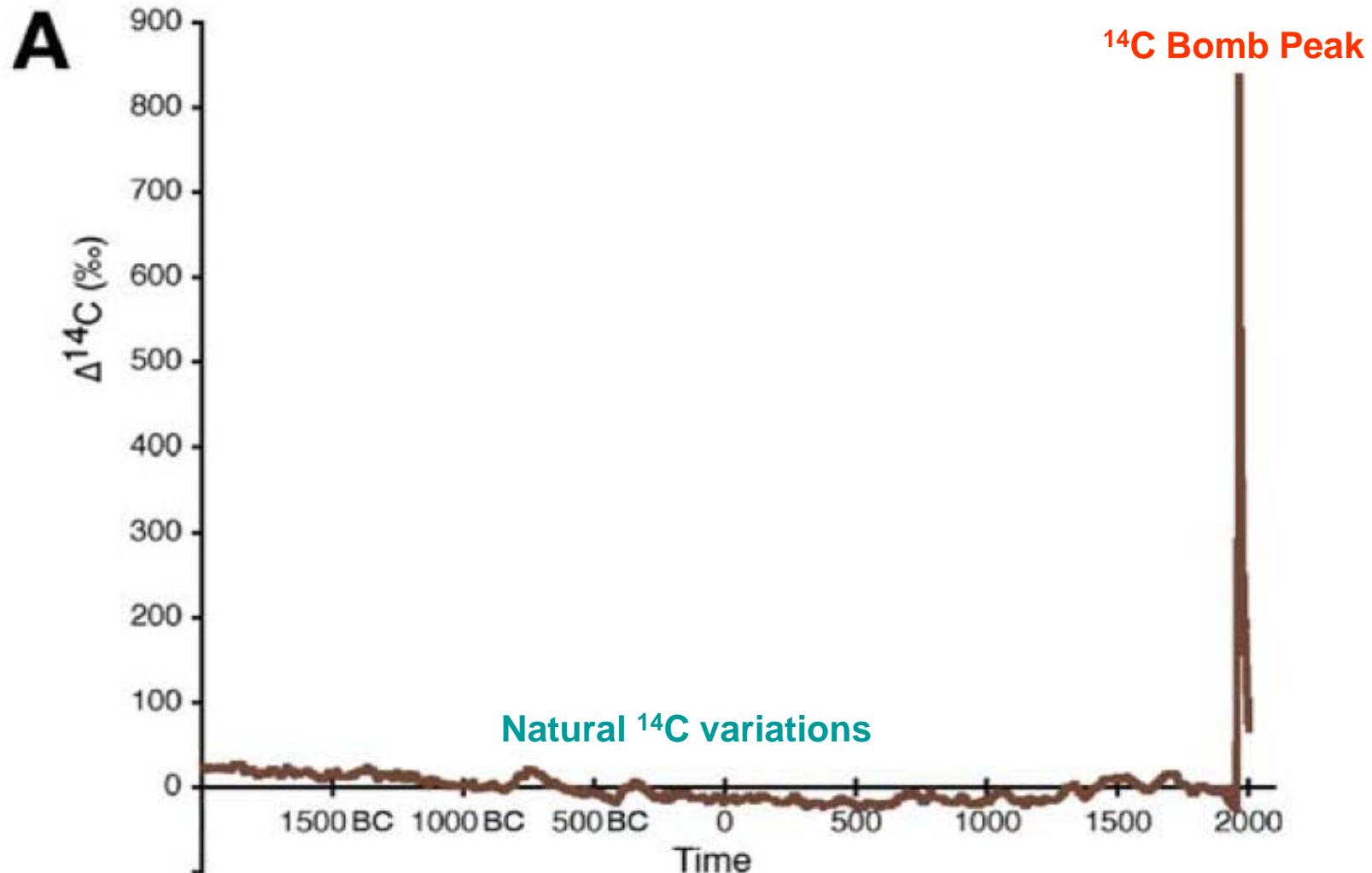
Equilibrium isotopic abundance of cosmogenic ^{14}C on Earth

^{12}C (0.99)
 stable

^{13}C (0.01)
 stable

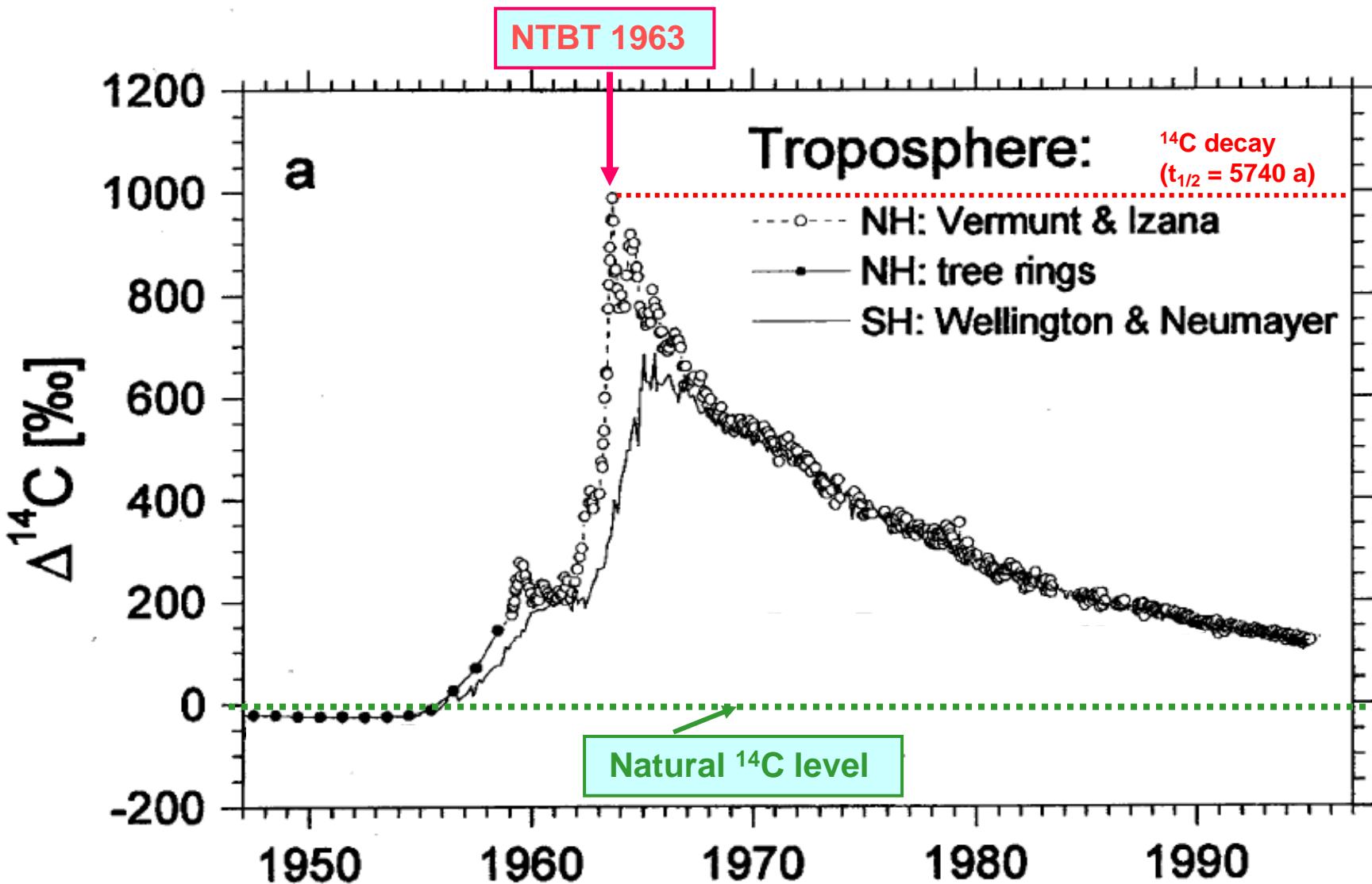
^{14}C (1.2×10^{-12})
 radioactive

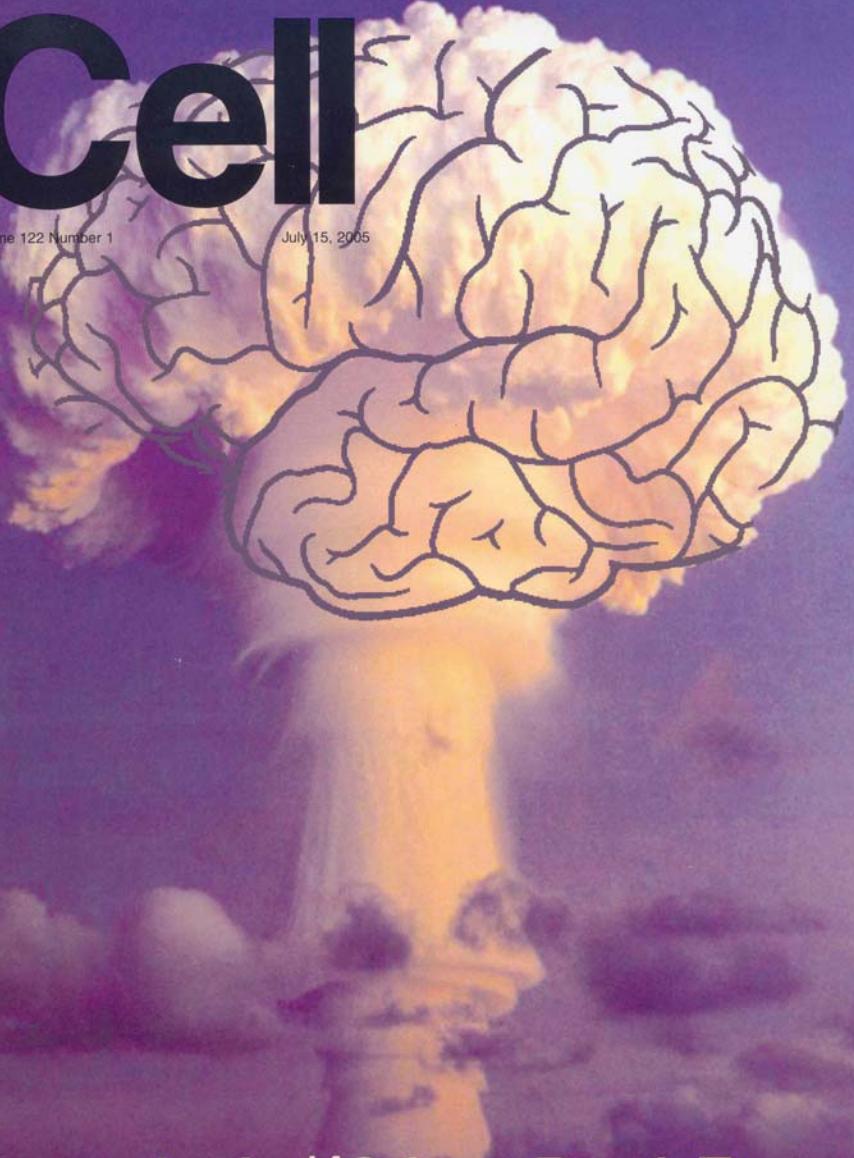
Variation of the ^{14}C content in atmospheric CO_2 during the last 4000 Years



Long-term observations of $\Delta^{14}\text{C}$ in atmospheric CO_2 in the northern and in the southern hemisphere

Levin and Hesshaimer, Univ. Heidelberg, Radiocarbon 42/1 (2000) 69





**Atmospheric ^{14}C from Bomb Tests
Aids in Dating Human Neurons**

Focus on miRNAs and RNAi

Retrospective Birth Dating of Cells in Humans

Kirsty L. Spalding et al.

*Department of Cell and Molecular
Biology, Medical Nobel Institute,
Karolinska Institute, Stockholm*

Cell, Vol. 122 (15 July 2005) 133-143

¹⁴C in genomic DNA reflects the age of cells

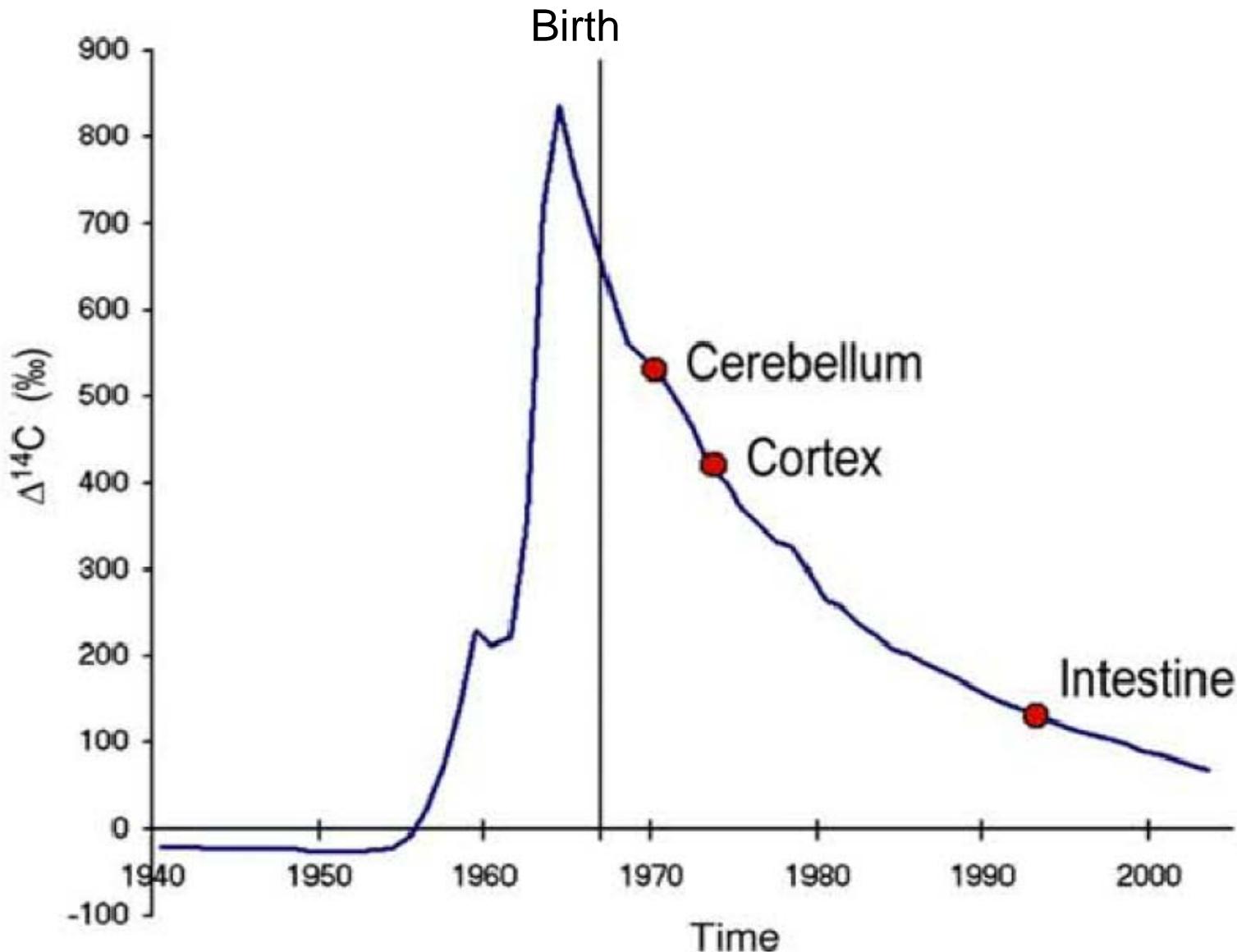
“Most molecules in a cell are in constant flux, with the unique exception of genomic DNA, which is not exchanged after a cell has gone through its last division.

The level of ¹⁴C integrated into genomic DNA should thus reflect the level in the atmosphere at any given point, and we hypothesized that determination of ¹⁴C levels in genomic DNA could be used to retrospectively establish the birth date of cells in the human body.”

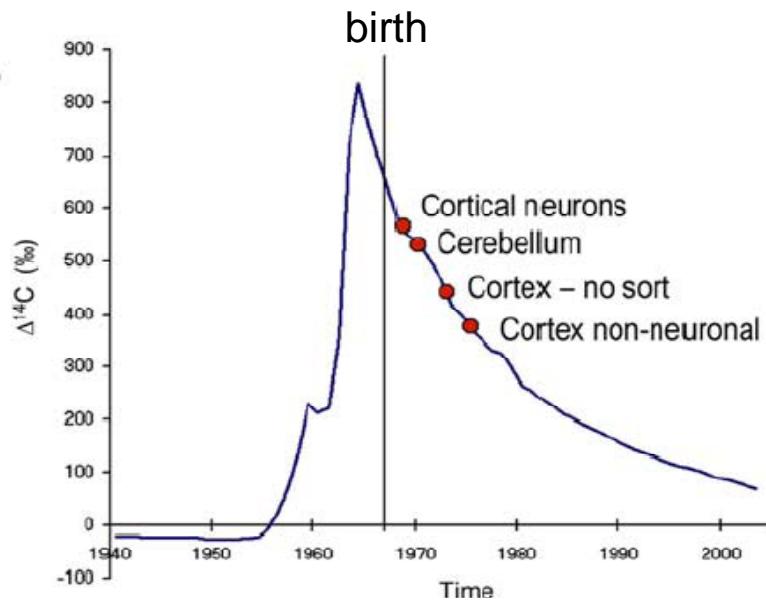
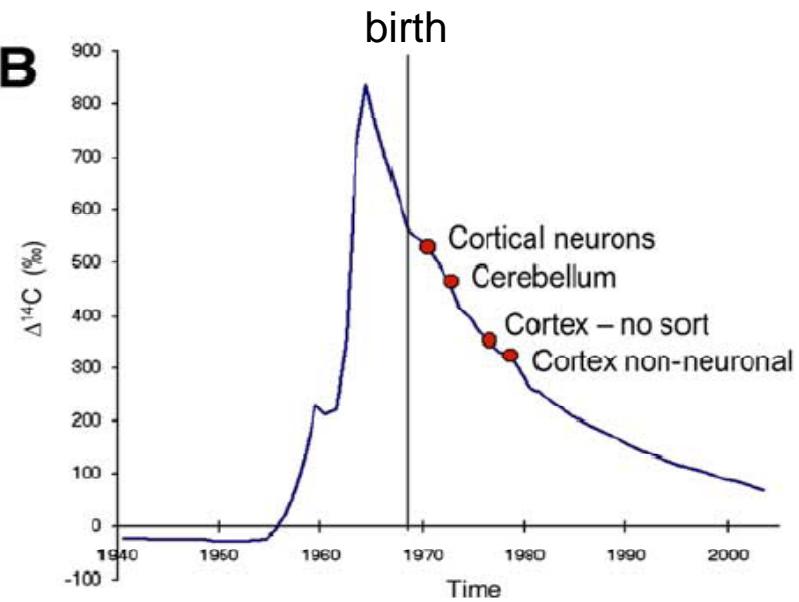
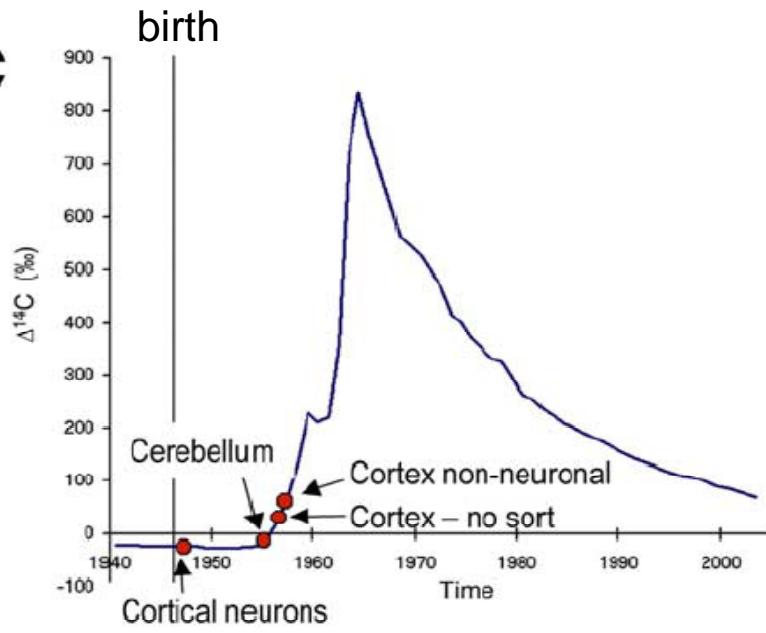
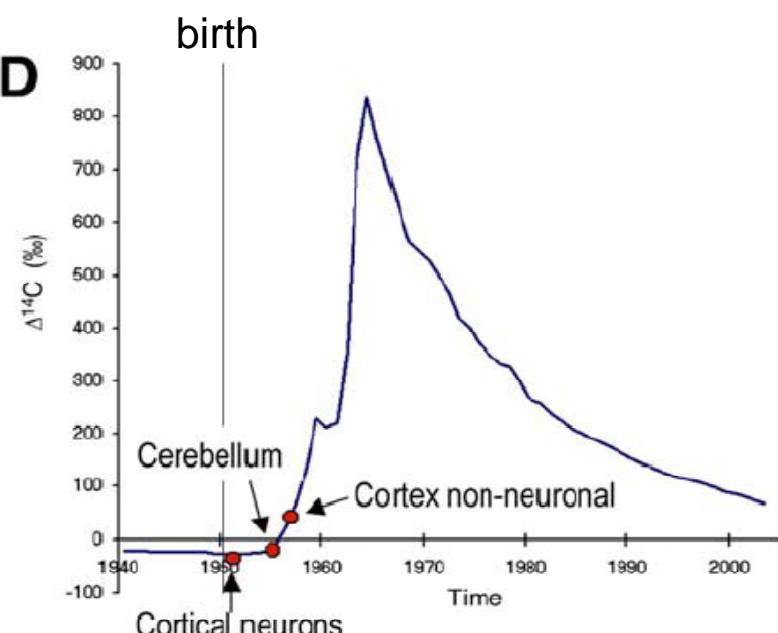
Some facts about human DNA and ^{14}C

(A physicist's view)

Basic composition of DNA:	Macromolecule with 3×10^9 basepairs
Chem. sum formula per bp:	$\text{C}_{20}\text{H}_{23}\text{N}_7\text{O}_{13}\text{P}_2$ and $\text{C}_{19}\text{H}_{22}\text{N}_8\text{O}_{13}\text{P}_2$
Molecular weight:	~630 daltons per base pair, total $\sim 1.9 \times 10^{12}$ Da
Mass of DNA per cell:	$2 \text{ DNA per cell} = 2 \times 3 \text{ pg} = 6 \text{ pg}$
Mass of carbon (40 wt% C):	2.4 pg
Total length of DNA per cell:	$2 \times 3 \times 10^9 \times (0.34 \text{ nm}) = 2 \text{ m}$
C atoms of DNA per cell:	$2 \times 3 \times 10^9 \times (20 \text{ C}) = \mathbf{1.2 \times 10^{11} \text{ C atoms}}$
$^{14}\text{C}/^{12}\text{C}$:	$\mathbf{1.2 \times 10^{-12}}$
DNA of 10 cells:	$\mathbf{\sim 1 \ ^{14}\text{C atom}}$
15 million cells:	1.5 million ^{14}C atoms
C from DNA of 15 million cells:	~36 μg C
Total ^{14}C detection efficiency:	~2% $\rightarrow \sim 30,000$ ^{14}C atoms detected



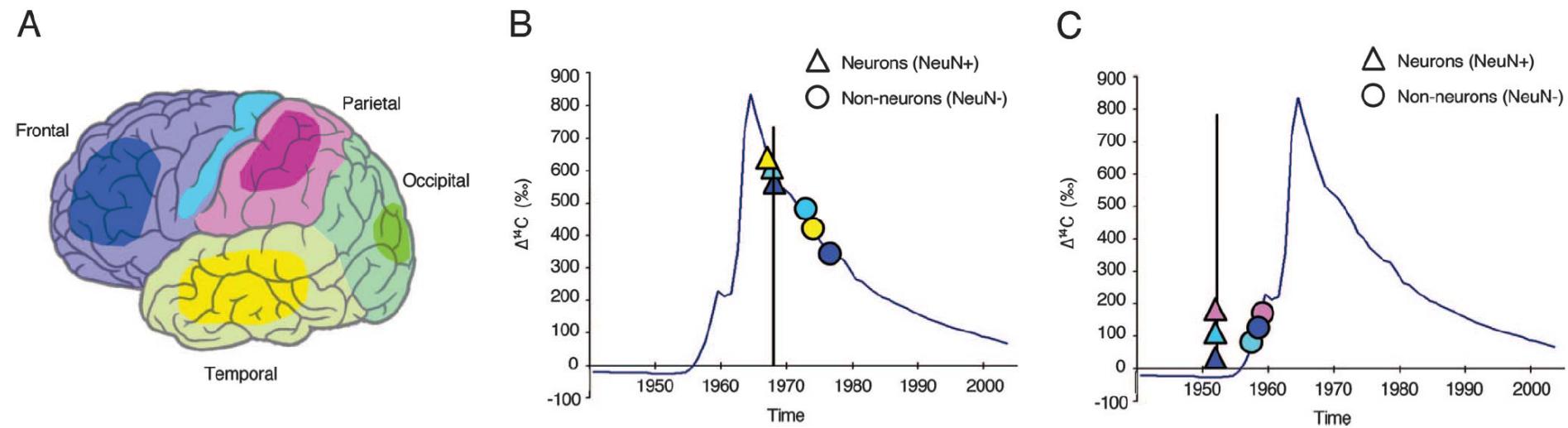
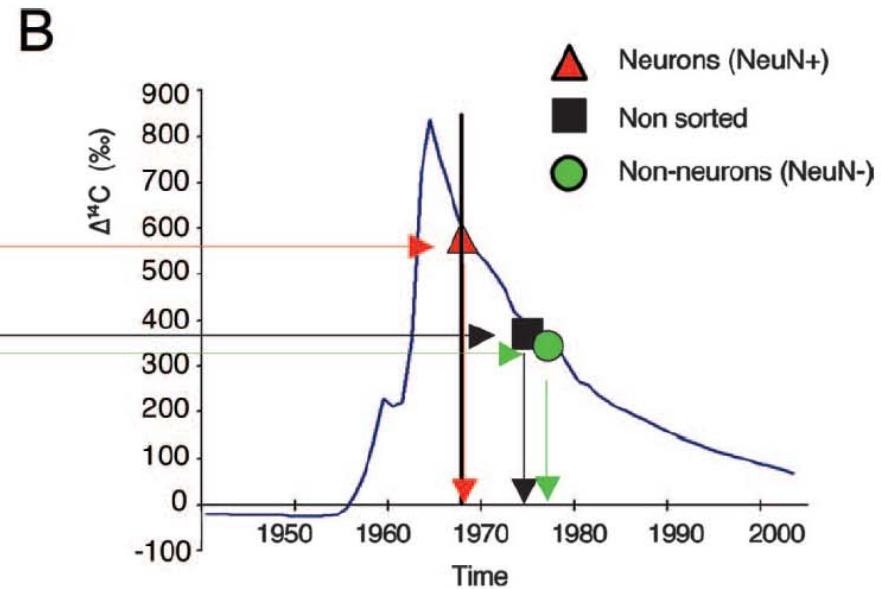
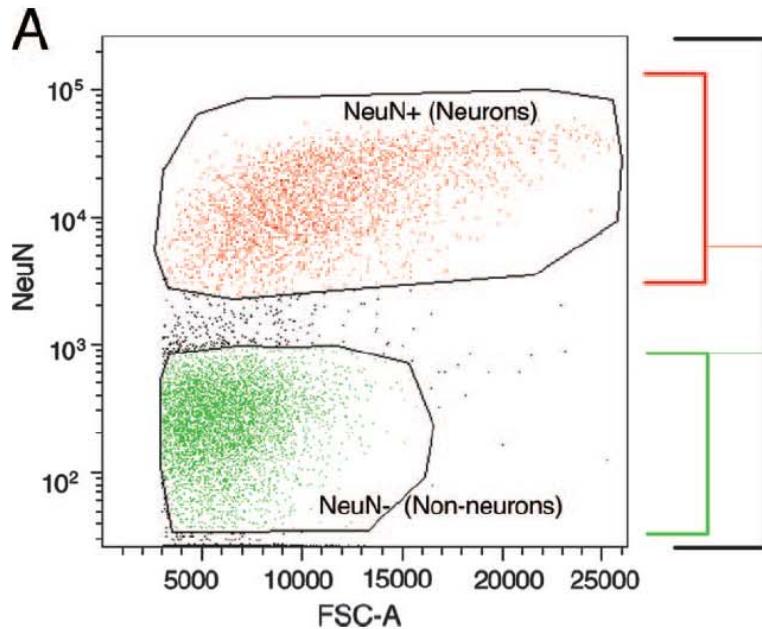
^{14}C dates from DNA extracted from the respective cells
Spalding et al., Cell 122 (2005) 133

A**B****C****D**

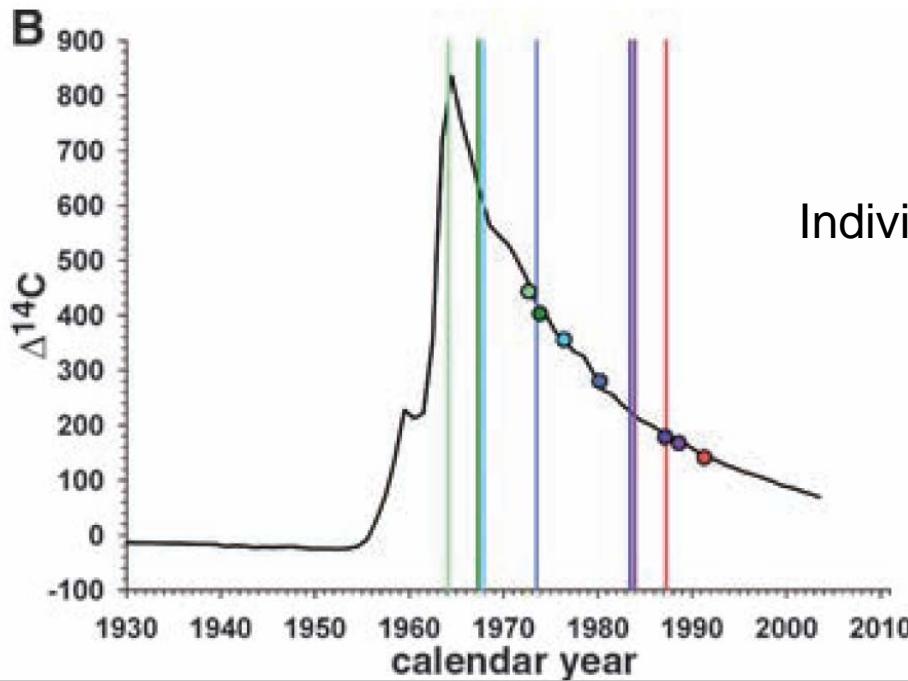
Cortical neurons and non-neuronal cells have different renewal rates
Spalding et al., Cell 122 (2005) 133

Topics studied with ^{14}C bomb peak dating of DNA

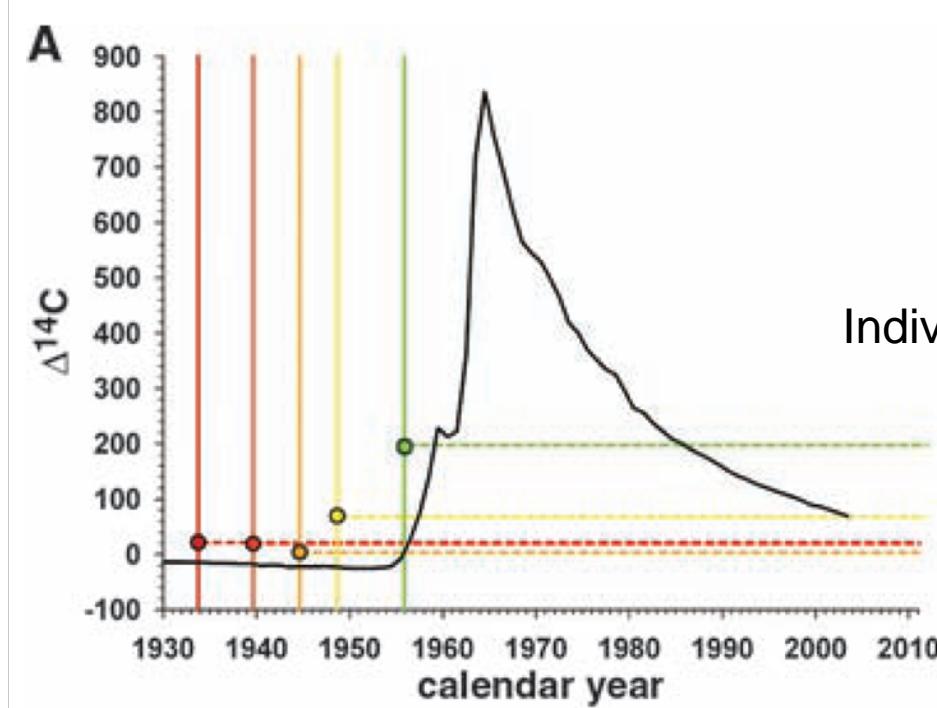
- **Reproduction of cells in different parts of the human brain**
K. Spalding et al., Cell 122 (2005) 133
- **Neocortical neurogenesis in humans is restricted to development**
R.D. Bahadurwaj et al., PNAS 103 (2006) 12564
- **Dynamics of fat cell turnover in humans**
K. Spalding et al., Nature 453 (2008) 783
- **Evidence for cardiomyocyte (heart cell) renewal in humans**
O. Bergmann et al., Science 324 (2009) 98



Determination of the age of neocortical neurons
Bhardwaj et al., PNAS 103 (2006) 12564



Individuals born after the ^{14}C bomb peak



Individuals born before the ^{14}C bomb peak

Evidence for renewal of human heart cells (cardiomyocytes)

Bergmann et al., *Science*
324 (2009) 98

The technical challenge

To develop ^{14}C AMS for μg -carbon samples. The DNA extracted from 10 human cell contains only one ^{14}C atom. For special sections of the brain little material is available.

Conclusion

A scenic view of Cape of Good Hope, showing its rugged, rocky coastline and surrounding mountains.

Cape of Good Hope

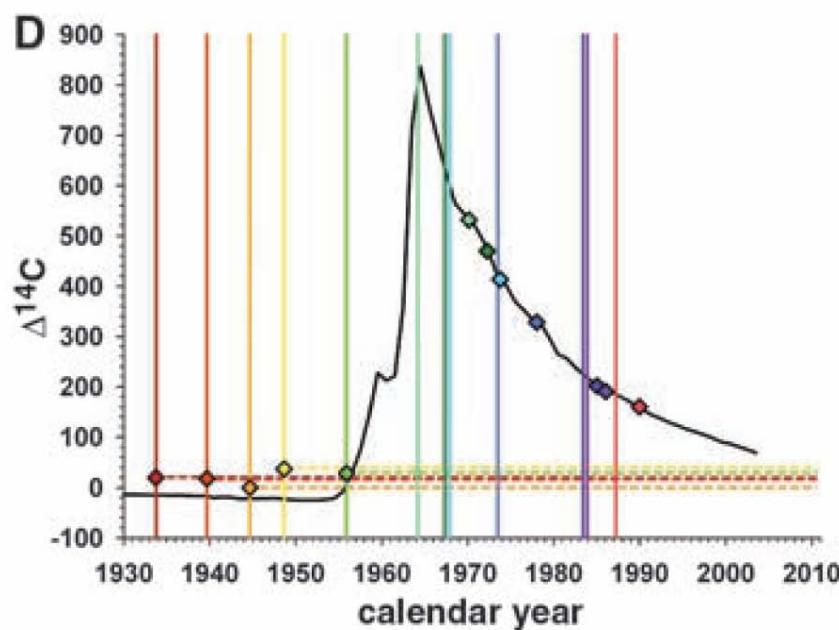
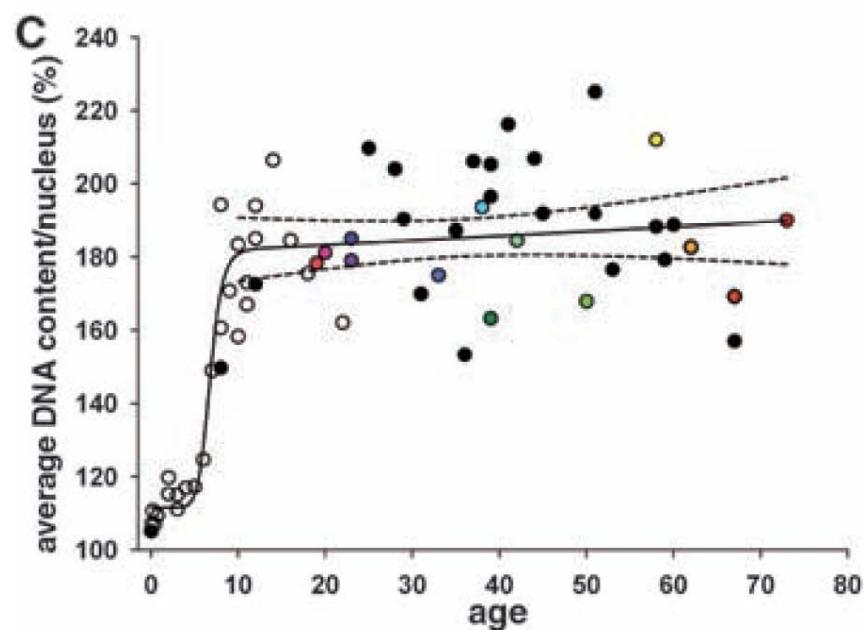
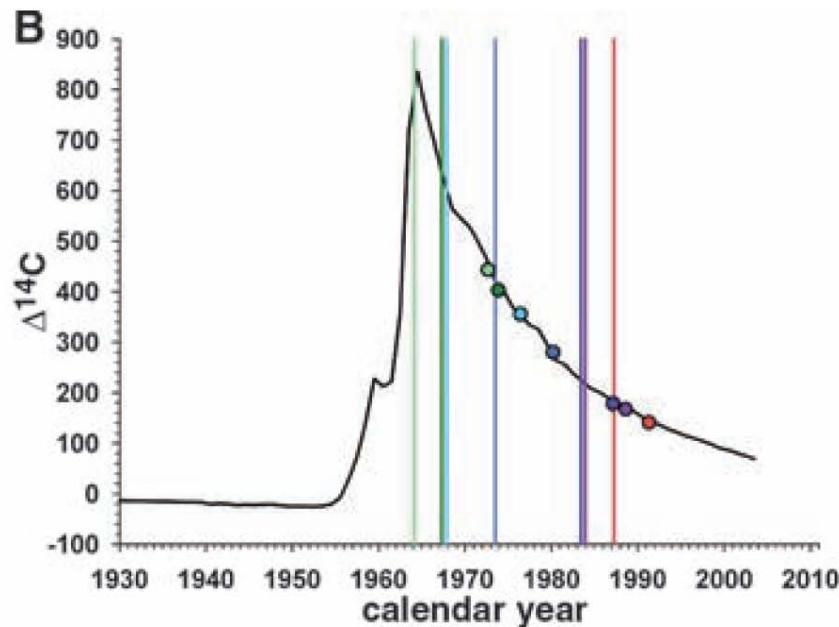
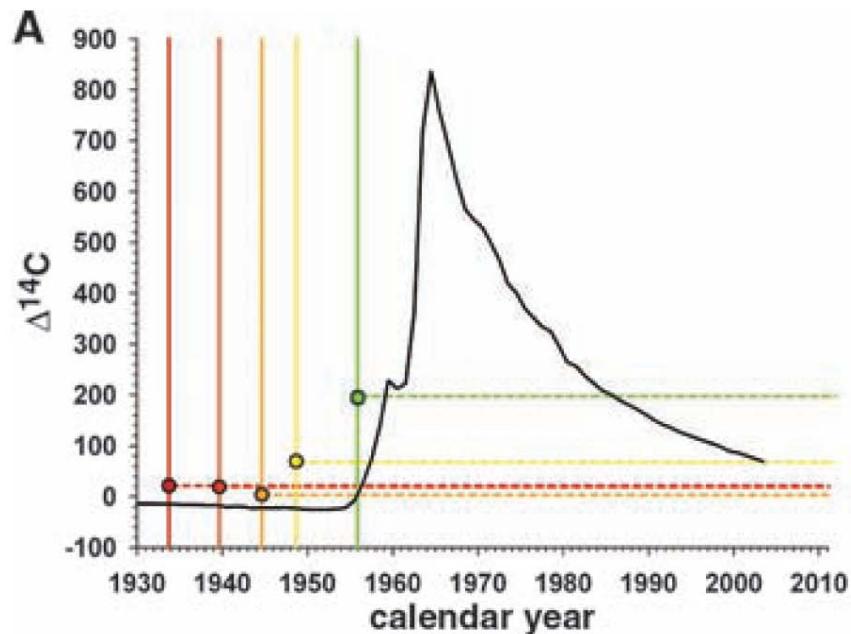


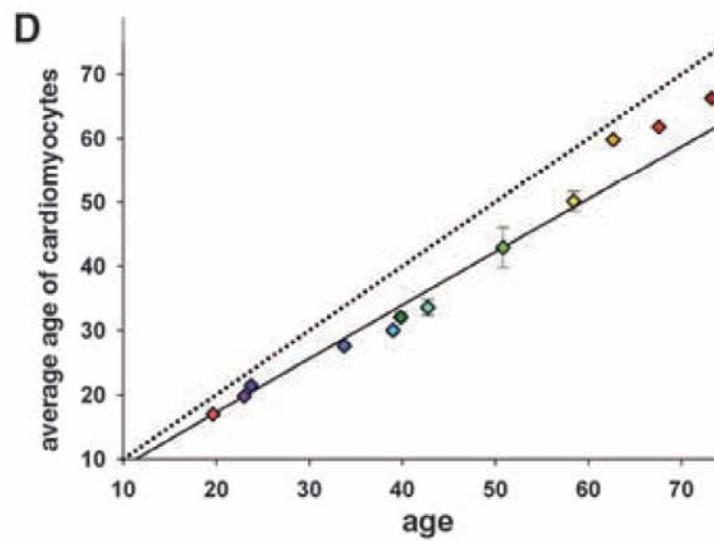
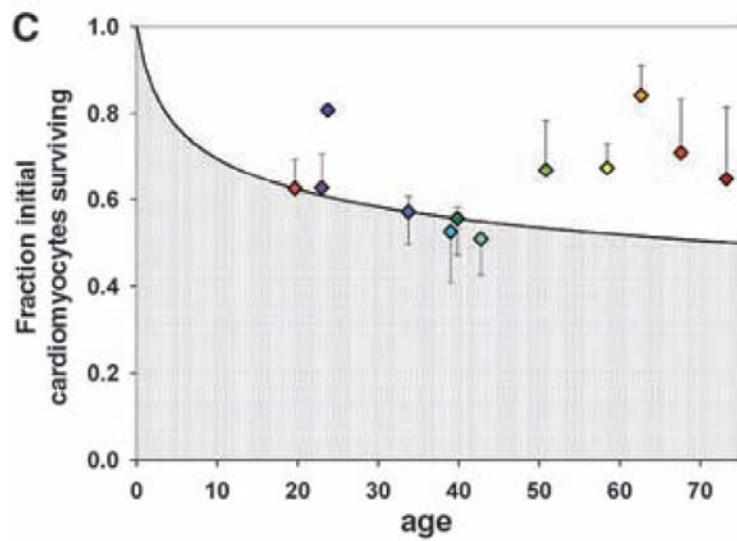
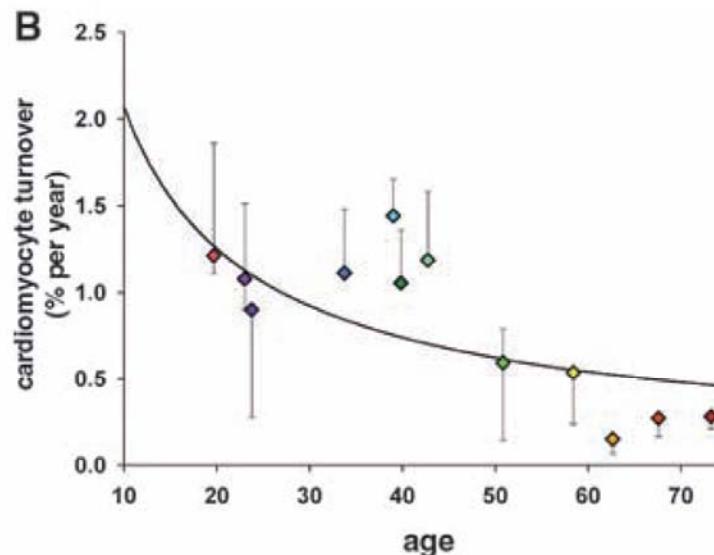
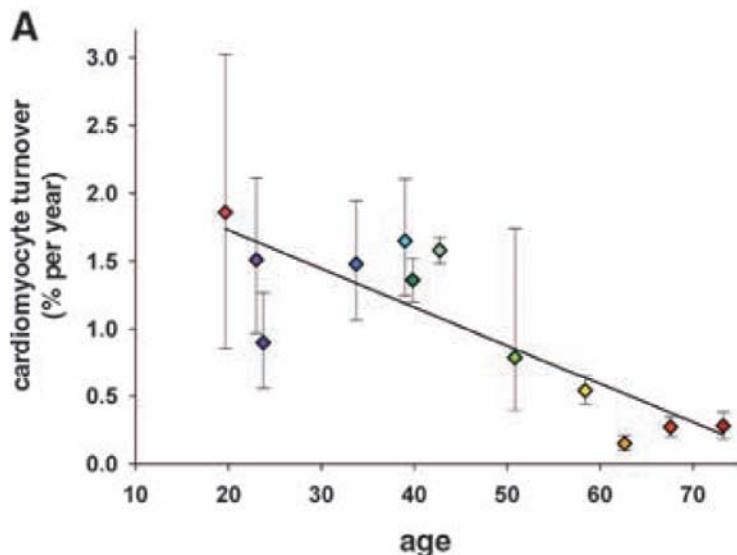
Photo: Georg Wolschin

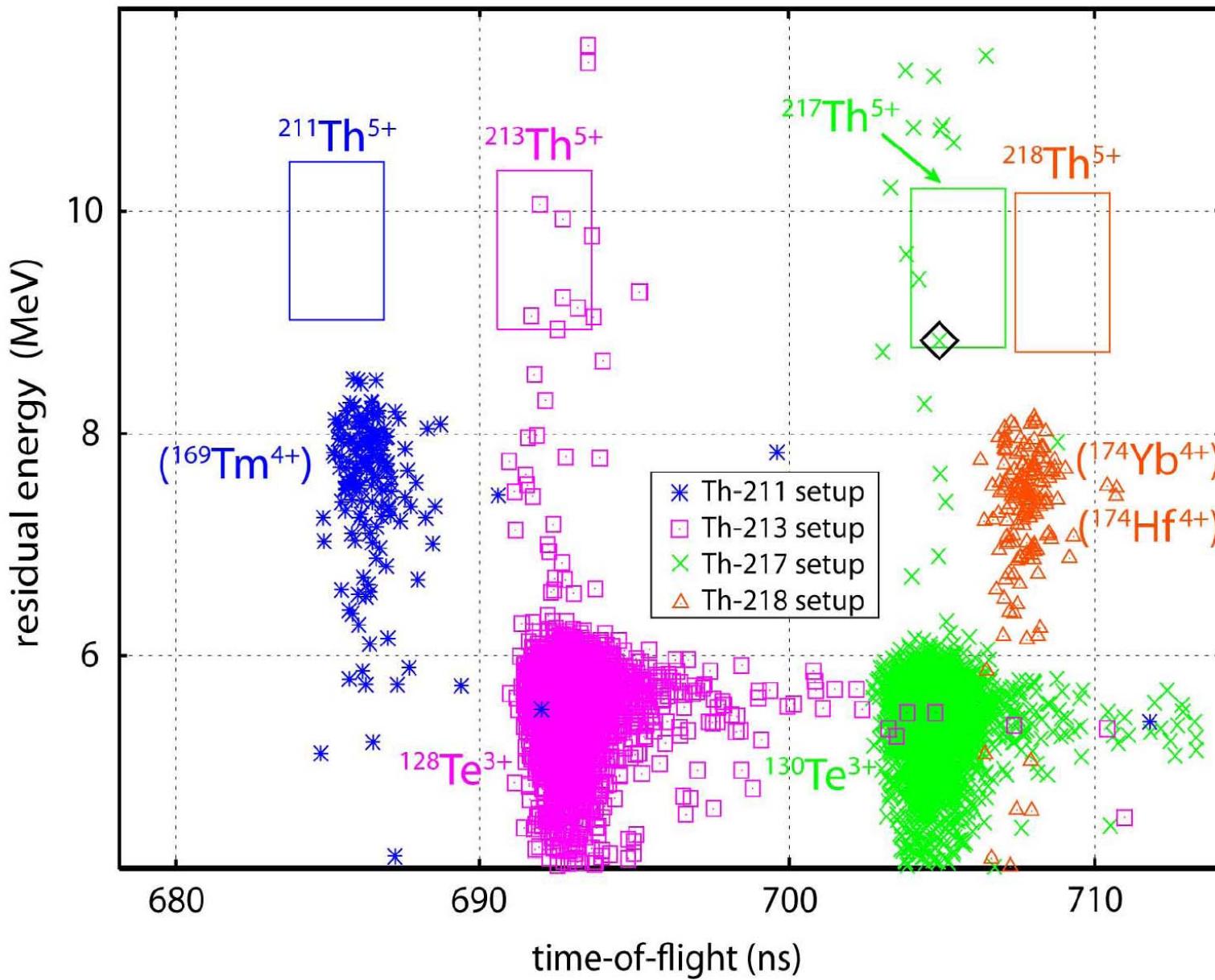


**It is better to be roughly right
than precisely wrong.**

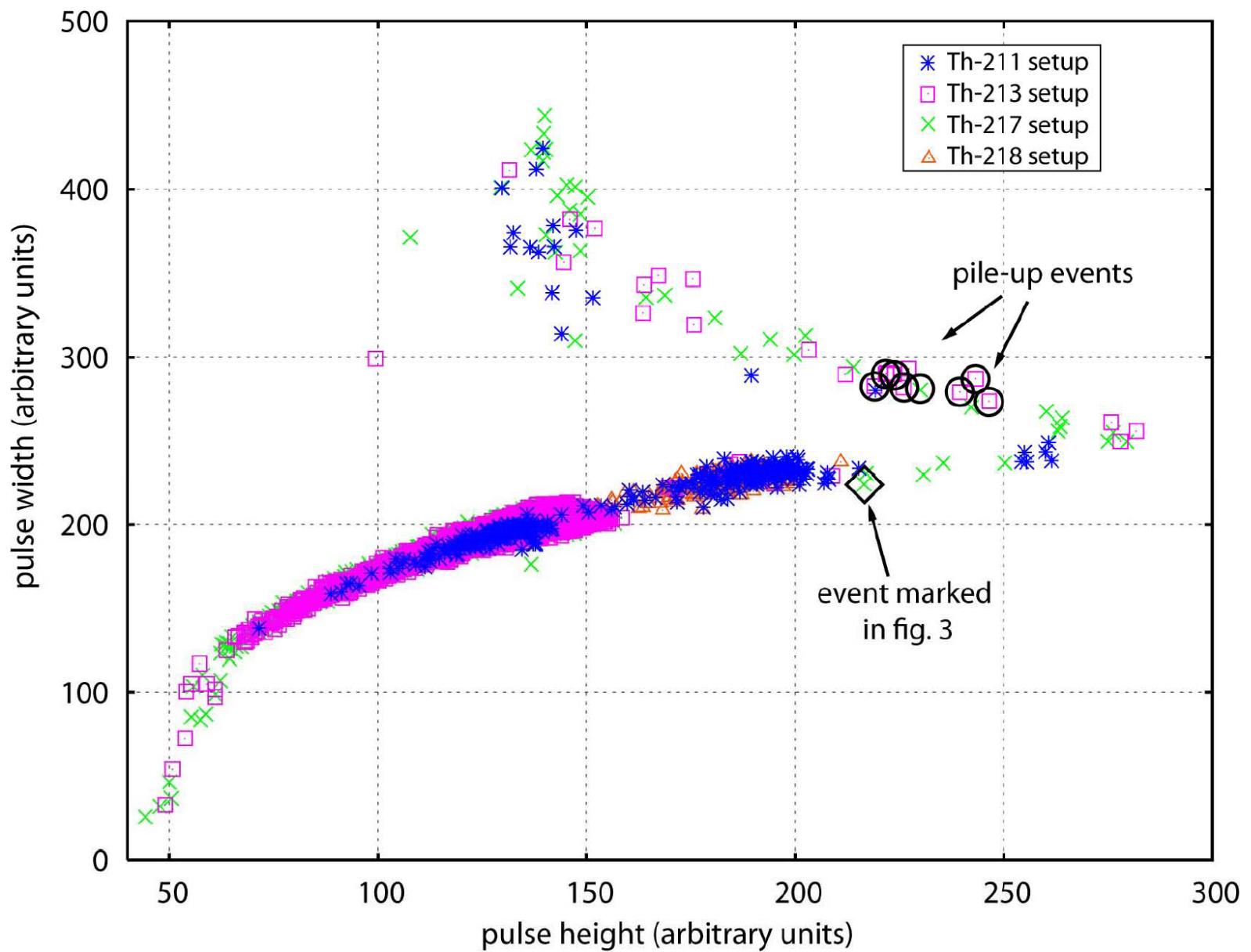
– Einstein







Search for neutron-deficient thorium isotopes



Identification of pile-up events

